

# Material Safety Data Sheet

## Acrylonitrile MSDS

### Section 1: Chemical Product and Company Identification

**Product Name:** Acrylonitrile

**Catalog Codes:** SLA2566

**CAS#:** 107-13-1

**RTECS:** Not available.

**TSCA:** TSCA 8(b) inventory: Acrylonitrile

**CI#:** Not available.

**Synonym:** Vinyl Cyanide; Propenitrile

**Chemical Name:** Acrylonitrile

**Chemical Formula:** C<sub>3</sub>H<sub>3</sub>N

**Contact Information:**

**Sciencelab.com, Inc.**

14025 Smith Rd.

Houston, Texas 77396

US Sales: **1-800-901-7247**

International Sales: **1-281-441-4400**

Order Online: [ScienceLab.com](http://ScienceLab.com)

**CHEMTREC (24HR Emergency Telephone), call:**

1-800-424-9300

**International CHEMTREC, call:** 1-703-527-3887

**For non-emergency assistance, call:** 1-281-441-4400

### Section 2: Composition and Information on Ingredients

**Composition:**

**Name CAS # % by Weight**

Acrylonitrile 1017-13-1 >99

**Toxicological Data on Ingredients:** Acrylonitrile: ORAL (LD50): Acute: 78 mg/kg [Rat]. 27 mg/kg [Mouse]. DERMAL (LD50):

Acute: 63 mg/kg [Rabbit]. VAPOR (LC50): Acute: 333 ppm 4 hours [Rat]. >90 ppm 4 hours [Monkey].

### Section 3: Hazards Identification

**Potential Acute Health Effects:**

Very hazardous in case of skin contact (irritant), of ingestion, of inhalation. Hazardous in case of skin contact (permeator), of eye contact (irritant). Severe over-exposure can result in death.

**Potential Chronic Health Effects:**

CARCINOGENIC EFFECTS: Classified 2B (Possible for human.) by IARC. MUTAGENIC EFFECTS: Mutagenic for

mammalian somatic cells. Mutagenic for bacteria and/or yeast. TERATOGENIC EFFECTS: Classified POSSIBLE for human.

DEVELOPMENTAL TOXICITY: Classified Reproductive system/toxin/female, Reproductive system/toxin/male [POSSIBLE].

The substance may be toxic to blood, kidneys, liver, cardiovascular system, central nervous system (CNS). Repeated or prolonged exposure to the substance can produce target organs damage. Repeated exposure to a highly toxic material may produce general deterioration of health by an accumulation in one or many human organs.

### Section 4: First Aid Measures

p. 2

**Eye Contact:**

Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. WARM water MUST be used. Get medical attention.

**Skin Contact:**

In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Cover the irritated skin with an emollient. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention immediately.

**Serious Skin Contact:**

Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.

**Inhalation:**

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.

**Serious Inhalation:**

Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband.

WARNING: It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled material is toxic, infectious or corrosive. Seek immediate medical attention.

**Ingestion:**

If swallowed, do not induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention immediately.

**Serious Ingestion:** Not available.

**Section 5: Fire and Explosion Data**

**Flammability of the Product:** Flammable.

**Auto-Ignition Temperature:** 481.11 °C (898 °F)

**Flash Points:** CLOSED CUP: -1.1111 °C (30 °F). OPEN CUP: 0 °C (32 °F).

**Flammable Limits:** LOWER: 3.1% UPPER: 17%

**Products of Combustion:** These products are carbon oxides (CO, CO<sub>2</sub>).

**Fire Hazards in Presence of Various Substances:**

Highly flammable in presence of open flames and sparks, of heat. Slightly flammable to flammable in presence of shocks.

**Explosion Hazards in Presence of Various Substances:**

Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available. Explosive in presence of oxidizing materials, of acids, of alkalis.

**Fire Fighting Media and Instructions:**

Flammable liquid, soluble or dispersed in water. SMALL FIRE: Use DRY chemical powder.

LARGE FIRE: Use alcohol foam, water spray or fog.

**Special Remarks on Fire Hazards:** Not available.

**Special Remarks on Explosion Hazards:**

In the presences of catalysts, or when the substance is confined, the polymerization rate may be accelerated leading to explosion. Acrylonitrile forms explosive mixtures with air based on its low flash point. It easily forms violently explosive polymerides when exposed to heat, light, strong bases, strong acids, strong oxidizers, azoisobutyronitrile, dibenzoyl peroxide, di-tert-butylperoxide, bromine or silver nitrate. Acrylonitrile may explosive reactions with benzyltrimethylammonium hydroxide + pyrrole. It may also have explosive reactions with tetrahydrocarbazole + benzyltrimethylammonium hydroxide.

**Section 6: Accidental Release Measures**

p. 3

**Small Spill:** Absorb with an inert material and put the spilled material in an appropriate waste disposal.

**Large Spill:**

Flammable liquid. Poisonous liquid. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not get water inside container. Do not touch spilled material. Use water spray to reduce vapors. Prevent entry into sewers, basements or confined areas; dike if needed. Call for assistance on disposal. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

**Section 7: Handling and Storage****Precautions:**

Keep locked up.. Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe gas/fumes/ vapor/spray. Wear suitable protective clothing. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidizing agents, metals, acids, alkalis.

**Storage:**

Light Sensitive. Store in light-resistant containers. Store in a segregated and approved area. Keep container in a cool, wellventilated area. Keep container tightly closed and sealed until ready for use. Avoid all possible sources of ignition (spark or flame).

**Section 8: Exposure Controls/Personal Protection****Engineering Controls:**

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-station location.

**Personal Protection:**

Splash goggles. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

**Personal Protection in Case of a Large Spill:**

Splash goggles. Full suit. Vapor respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

**Exposure Limits:**

TWA: 2 (ppm) from ACGIH (TLV) [United States] TWA: 4.3 (mg/m3) TWA: 1 CEIL: 10 from NIOSH TWA: 2 STEL: 10 (ppm) from OSHA (PEL) [United States] TWA: 2 (ppm) [United Kingdom (UK)] TWA: 4.3 (mg/m3) [United Kingdom (UK)] TWA: 2 STEL: 4 (ppm) [Canada] TWA: 4.3 STEL: 8.6 (mg/m3) [Canada]<sup>3</sup> Consult local authorities for acceptable exposure limits.

**Section 9: Physical and Chemical Properties**

**Physical state and appearance:** Liquid.

**Odor:** mild Peach kernels. (Slight.)

**Taste:** Not available.

**Molecular Weight:** 53.06 g/mole

**Color:** Clear Colorless.

**pH (1% soln/water):** Not available.

**Boiling Point:** 77.3°C (171.1°F)

**Melting Point:** -82°C (-115.6°F)

**Critical Temperature:** 262.78 °C (505 °F)

**Specific Gravity:** 0.806 (Water = 1)

**Vapor Pressure:** 11.1 kPa (@ 20 °C)

**Vapor Density:** 1.8 (Air = 1)

**Volatility:** Not available.

**Odor Threshold:** Not available.

**Water/Oil Dist. Coeff.:** The product is more soluble in oil;  $\log(\text{oil/water}) = 0.3$

**Ionicity (in Water):** Not available.

**Dispersion Properties:** See solubility in water, diethyl ether, acetone.

**Solubility:**

Soluble in diethyl ether, acetone. Very slightly soluble in cold water, hot water. Soluble in all common organic substances and

Isopropyl alcohol. Soluble in benzene and alcohol

**Section 10: Stability and Reactivity Data**

**Stability:** The product is stable.

**Instability Temperature:** Not available.

**Conditions of Instability:**

Heat, ignition sources, light, loss of inhibitor. Polymerization may occur, especially when exposed to visible light or in the absence of oxygen

**Incompatibility with various substances:**

Highly reactive with oxidizing agents, acids, alkalis. Reactive with metals.

**Corrosivity:** Highly corrosive in presence of aluminum, of copper.

**Special Remarks on Reactivity:**

Light Sensitive. Incompatible with strong oxidizers, strong acids (nitric acid, sulfuric acid, chlorosulfonic acid), strong bases

(potassium hydroxide, sodium hydroxide), amines, 2-aminoethanol, bromine, ethylene diamine, oleum. Unless inhibited

(usually with methylhydroquinone), this material may spontaneously polymerize, or it may spontaneously polymerization

under certain conditions. Polymerization reactions are usually highly exothermic. Small amounts of acids (nitric or sulfuric)

may neutralize the ammonia used to inhibit acrylonitrile and create uninhibited, unstable

acrylonitrile. Strong bases will cause

acrylonitrile to violently polymerize. It may spontaneously polymerize when heated, or exposed to light.

**Special Remarks on Corrosivity:** It attacks copper, copper alloys, and aluminum in high concentrations.

**Polymerization:** Yes.

**Section 11: Toxicological Information**

**Routes of Entry:** Absorbed through skin. Dermal contact. Eye contact. Inhalation. Ingestion.

**Toxicity to Animals:**

WARNING: THE LC50 VALUES HEREUNDER ARE ESTIMATED ON THE BASIS OF A 4-HOUR EXPOSURE. Acute oral

toxicity (LD50): 27 mg/kg [Mouse]. Acute dermal toxicity (LD50): 63 mg/kg [Rabbit]. Acute toxicity of the vapor (LC50): >90 4

hours [Monkey].

**Chronic Effects on Humans:**

CARCINOGENIC EFFECTS: Classified 2B (Possible for human.) by IARC. MUTAGENIC EFFECTS: Mutagenic for

mammalian somatic cells. Mutagenic for bacteria and/or yeast. TERATOGENIC EFFECTS: Classified POSSIBLE for human.

DEVELOPMENTAL TOXICITY: Classified Reproductive system/toxin/female, Reproductive system/toxin/male [POSSIBLE].

May cause damage to the following organs: blood, kidneys, liver, cardiovascular system, central nervous system (CNS).

p. 5

**Other Toxic Effects on Humans:**

Very hazardous in case of skin contact (irritant), of ingestion, of inhalation. Hazardous in case of skin contact (permeator).

**Special Remarks on Toxicity to Animals:** Not available.

**Special Remarks on Chronic Effects on Humans:**

May cause adverse reproductive effects (fetotoxicity, maternal effects on fertility, paternal effects on fertility). May affect

genetic material (mutagenic). May cause cancer (tumorigenic) based on animal data. It is a suspect human carcinogen. May

cause birth defects (musculoskeletal, central nervous system, cardiovascular)

**Special Remarks on other Toxic Effects on Humans:**

Acute Potential Health Effects: Skin: Causes severe skin irritation. Prolonged skin contact may cause formation of large

vesicles after a latent period of several hours. The burns resemble second degree thermal burns, but with little pain or

inflammation. May be absorbed through skin. May be harmful if absorbed through skin. May affect behavior/central nervous

system, respiration and gastrointestinal tract is absorbed through skin. Eyes: Causes moderate eye irritation. Lachrymator.

Inhalation: May be harmful if inhaled. Inhalation of high concentrations may affect behavior/central nervous system with

symptoms including central nervous system depression, seizures, weakness in the limbs, dizziness, impaired judgement,

irritability, apprehension, weakness, lightheadedness, headache, anxiety, agitation, stupor, seizures, ataxia, confusion, coma.

May also affect cardiovascular system (palpitations, arrhythmias, cardiac conduction defects, rapid heartbeat), respiration

(hyperventilation, dyspnea), gastrointestinal system (nausea, vomiting). Inhalation may cause cyanosis (a bluish discoloration

of the skin due to deficient oxygenation of the blood). Ingestion: Harmful if swallowed. Causes digestive tract irritation with

nausea, and vomiting. May affect behavior/central nervous system, and respiration with symptoms similar to inhalation.

Chronic Potential Health Effects: Repeated contact by inhalation or ingestion may affect the liver (jaundice), urinary system

(kidneys), metabolism. Repeated contact by ingestion may also affect the blood (anemia).

## **Section 12: Ecological Information**

**Ecotoxicity:** Not available.

**BOD5 and COD:** Not available.

**Products of Biodegradation:**

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

**Toxicity of the Products of Biodegradation:** The products of degradation are less toxic than the product itself.

**Special Remarks on the Products of Biodegradation:** Not available.

## **Section 13: Disposal Considerations**

**Waste Disposal:**

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

## **Section 14: Transport Information**

**DOT Classification:**

CLASS 3: Flammable liquid. CLASS 6.1: Poisonous material.

**Identification:** : Acrylonitrile, inhibited UNNA: 1093 PG: I

**Special Provisions for Transport:** Not available.

## **Section 15: Other Regulatory Information**

**Federal and State Regulations:**

California prop. 65: This product contains the following ingredients for which the State of California has found to cause cancer,

birth defects or other reproductive harm, which would require a warning under the statute:  
Acrylonitrile California prop. 65:

p. 6

This product contains the following ingredients for which the State of California has found to cause cancer which would

require a warning under the statute: Acrylonitrile Connecticut hazardous material survey.:

Acrylonitrile Illinois toxic substances

disclosure to employee act: Acrylonitrile Illinois chemical safety act: Acrylonitrile

**Other Regulations:**

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).

EINECS: This product is on the

European Inventory of Existing Commercial Chemical Substances.

**Other Classifications:**

**WHMIS (Canada):** Not controlled under WHMIS (Canada).

**DSCL (EEC):**

R11- Highly flammable. R16- Explosive when mixed with oxidizing substances. R23/24/25- Toxic by inhalation, in contact

with skin and if swallowed. R36/38- Irritating to eyes and skin. R40- Possible risks of irreversible effects. R62- Possible risk

of impaired fertility. R63- Possible risk of harm to the unborn child. S1/2- Keep locked up and out of the reach of children.

S36/37- Wear suitable protective clothing and gloves. S45- In case of accident or if you feel unwell, seek medical advice

immediately (show the label where possible). S46- If swallowed, seek medical advice immediately and show this container or label.

**HMIS (U.S.A.):**

**Health Hazard:** 3

**Fire Hazard:** 3

**Reactivity:** 1

**Personal Protection:** h

**National Fire Protection Association (U.S.A.):**

**Health:** 4

**Flammability:** 3

**Reactivity:** 2

**Specific hazard:**

**Protective Equipment:**

Gloves. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Splash goggles.

**Section 16: Other Information**

**References:** Not available.

**Other Special Considerations:** Not available.

**Created:** 10/09/2005 03:37 PM

**Last Updated:** 11/01/2010 12:00 PM

*The information above is believed to be accurate and represents the best information currently available to us. However, we*

*make no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume*

*no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for*

*their particular purposes. In no event shall ScienceLab.com be liable for any claims, losses, or damages of any third party or for*

*lost profits or any special, indirect, incidental, consequential or exemplary damages, howsoever arising, even if ScienceLab.com*

*has been advised of the possibility of such damages.*

## Specifications for Acrylonitrile

Color APHA, (Max.) 15  
Moisture, (Wt. % Max) 0.5  
Specific Gravity @ 20 °C 0.799 – 0.802  
Acidity as CH<sub>3</sub>COOH (Wt. % Max) Report  
Distillation Range, °C 74.5- 79.0  
Iron as Fe, (Wt. ppm Max) Report  
Copper, (Wt. ppm Max.) Report  
PH ( 5 % Aq. Solution) 6.0 – 7.5  
Refractive Index @ 25 °C 1.3882 – 1.3895  
Inhibitor (MEHQ, Wt. ppm) 35 – 45  
Non volatile matter, ( ppm Max) 50  
Oxidation Stability, (Minutes. Min) 240  
Aldehyde as CH<sub>3</sub>CHO, (ppm Max) Report  
Acroline, (ppm Max) Report  
Acetone, (ppm Max) Report  
Acetonitrile, (ppm Max.) Report  
Purity, (Wt. % Min) 99.4

### Uses:

Polymer In mfr of acrylic fibres  
Coating As resin  
Adhesives As resin  
Pharma In mfr of antioxidants  
Agrochemicals as pesticide fumigant