PIPERAZINE

PRODUCT IDENTIFICATIO	NC	
CAS NO.	110-85-0 ,	
EINECS NO.	203-808-3	
FORMULA	NHCH ₂ CH ₂ NHCH ₂ CH ₂	
MOL WT.	86.14 H	
H.S. CODE	2933.59.9500	
TOXICITY	Oral, rat LD50: 1900 mg/kg	
SYNONYMS	Diethylenediamine; Diethyleneimine; Dispermine;	
Antiren; Hexahydropyro	zine; Piperazidine; Pipersol; Pyrazine hexahydride; Uvilon; 1,4-	
Diazacyclohexane; 1,4	-Piperazine; 1,4-Diethylenediamine; Piperazin (German); N,N-	
Diethylenediamine; He	xahydro-1,4-diazine; Piperazidine; Pyrazine nexanyarate; Piperazin	
(German); Piperazina (3 854880-15-2, 861800-35-	-3	
SMILES	CICNCCNI	
CLASSIFICATION	Piperazine, Anthelmintic	
EXTRA NOTES		
PHYSICAL AND CHEMICAL PROPERTIES		
PHYSICAL STATE	White flake Ammonia like odor	
MELTING POINT	108 C	
BOILING POINT	145 - 146 C	
SPECIFIC GRAVITY	1.1	
SOLUBILITY IN WATER	150 g/l at 20 C	
SOLVENT SOLUBILITY	Freely soluble in glycerol	
	3.0	
AUTOIGNITION	340 C 0.72 (Dissectation Constant at 05 C)	
	1.50E+00 (Octonel water)	
	0.16 (mmHa at 25 C)	
	$2.00E_{00}$ (at mm m3/mala at 25 C)	
	2.20E-09 (difficility) mole di 2.5 C)	
OH RAIE CONSTANT	1.69E-10 (cm3/molecule-sec at 25 C Atmospheric)	
	Health: 2 ; Flammability: 2 ; Reactivity: 0	
STARII ITY	Stable under ordinary conditions	
EXTERNAL LINKS & GENE		
Piperazine is a six-sided	organic ring compound containing two opposing nitrogen atoms (see	
image). The piperazines are a broad class of chemical compounds, all of which contain a		
piperazine functional group. Piperazines were originally named because of their chemical similarity		
with piperidine, a constiuent of piperine in the black pepper plant (Piper nigrum). This has led to the		
erroneous belief that piperazines are naturally derived from black pepper. In reality, no piperazines		
occur naturally; they are usually artiticially synthesized by reacting alcoholic ammonia with 1,2-		
aichioroethane. Many piperazines are successful drugs. Notable piperazine drugs include:		
DZF (lectedilonal alug)		

Dramamine (motion sickness drug) Viagra (impotence drug) A large number of piperazine compounds have anthelmintic (anti-parasitic) action. Their mode of action is generally by paralysing parasites, which allows the host body to easily remove or expel the invading organism. Piperazine hydrate and piperazine citrate are the main anthelminthic piperazines. These drugs are often referred to simply as "piperazine" which may cause confusion between the specific anthelmintic drugs and the entire class of piperazine-containing compounds. Piperazines are also used in the manufacture of plastics, resins, pesticides, and other industrial materials.

<u>Local:</u>

Piperazine is a heterocyclic compound containing four carbon atoms and two of nitrogen at 1,4 position (als called 1,4-hexahydropyrazine). It is a deliquescent crystalline compound melting at 105 C; soluble in water, alcohol, glycerol, and glycols. It is used as a main ingredient of anthelmintics used to treat intestinal roundworms (ascariasis) infection in human and poultry and to treat pinworms (enterobiasis; oxyuriasis) by altering cell membrane permeability and causing hyperpolarization of the membrane. Piperazine is a main moiety in psychoactive drugs. Certain piperazine derivatives are suspected of ecstasy substitutes. Benzylpiperazine is bannaed in many countries. Benzylpiperazine has been used as a anthelmintic (antiparastic effect). Nitrogen in piperazine ring plays an important role in biological research and drug manufacturing industry including the preparation of anthelmintic, antiallergenic, antibacterial, antihistamic, antiemetic and antimigraine agents. The piperazine ring and piperazine derivatives are important cyclic components in industrial field as raw materials for hardener of epoxy resins, corrosion inhibitors, insecticides, accelerators for rubber, urethane catalysts and antioxidants.

SALES SPECIFICATION	
APPEARANCE	White flake
PURITY	99.5% min
WATER	0.5% max
INDIVIDUAL	
IMPURITY	0.5% max
TRANSPORTATION	
PACKING	25kgs in bag
HAZARD CLASS	8 (Packing Group: III)
UN NO.	2579
SAFETY INFORMATION	
GHS	
SIGNAL WORD	Warning
PICTOGRAMS	$\mathbf{A} \mathbf{A} \mathbf{A}$
HAZARD STATEMENTS	H314: Causes severe skin burns and eye damage
	H317: May cause an allergic skin reaction
	H334: May cause allergy or asthma symptoms or breathing difficulties if inhaled
	H361fd: Suspected of damaging fertility. Suspected of damaging the
	H228. Flammable solid
PRECAUTIONARY	P301 + P330 + P331: IF SWALLOWED: rinse mouth. Do NOT induce vomiting
STATEMENTS	P305 + P351 + P338: IF IN EYES: Rinse cautiously with water for several
	minutes. Remove contact lenses, if present and easy to do. Continue rinsing
	P261: Avoid breathing dust/fume/gas/mist/vapors/spray
	P304 + P341: IF INHALED: If breathing is difficult, remove to fresh air and keep
	at rest in a position comfortable for breathing

ec directives	P342 + P311: If experiencing respiratory symptoms: Call a POISON CENTER or doctor/physician P280: Wear eye protection/face protection P273: Avoid release to the environment
HAZARD CODES	F Highly flammable
RISK PHRASES	 Highly flammable. Causes burns. May cause sensitisation by inhalation and skin contact. Possible risk of impaired fertility.
SAFETY PHRASES	 63: Possible risk of harm to the unborn child. 22: Do not breathe dust. 26: In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. 36/37/39: Wear suitable protective clothing, gloves and eye/face protection. 45; In case of accident or if you feel unwell, seek medical advice immediately (show label where possible) 61: Avoid release to the environment. Refer to special instructions safety data sheet.