FORMAMIDE

PRODUCT IDENTIFICATION	
CAS NO.	75-12-7 버 🖉
EINECS NO.	200-842-0
FORMULA	HCONH ₂ I NH ₂
MOL WT.	45.04
h.s. code	
TOXICITY	Oral rat LD50: 5577 mg/kg
synonyms	Methanamide; Carbamaldehyde; Formimidic Acid;
DERIVATION	
CLASSIFICATION	
PHYSICAL AND CHEMICAL PROPERTIES	
PHYSICAL STATE	Clear liquid with ammonia odor.
MELTING POINT	2.5 C
BOILING POINT	210 C (Decomposes)
SPECIFIC GRAVITY	1.13 - 1.14
SOLUBILITY IN WATER	miscible
рН	7
VAPOR DENSITY	1.6
AUTOIGNITION	
NFPA RATINGS	Health: 2; Flammability: 1; Reactivity: 0
REFRACTIVE INDEX	
FLASH POINT	154 C
STABILITY	Stable under ordinary conditions
GENERAL DESCRIPTION & APPLICATIONS	
Formamide, HCONH ₂ is an amide derived from formic acid, the simplest and lowest mole weight	
carboxylic acid in which a single hydrogen atom is attached to the carboxyl group (HCOOH).	
Amide is a group of organic chemicals with the general formula RCONH2 in which a carbon atom is	
attached to oxygen in solid bond and also attached to an hydroxyl group, where R is hydrogen in	
formamide. Formamide is an transparent hygroscopic oily liquid with an ammonia odor. It is soluble	
in water and alcohol; melting point 2.5 C, boiling 210 C. Formamide is obtained by heating ethyl	
formate with ammonia or by heating ammonium formate with urea. Formamiae is used in the	
synthesis of medicine, pesticide, animal give and other organic compound. It is used as a solvent	
which ionizes water-insol	uble compounds. It is used as a softening agent for paper and fiber. It is
sales specification	
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Hazara Symbols: 1, Risk Phrases: 61, Satety Phrases: 45-53	