4-CYANOPYRIDINE

PRODUCT IDENTIFICATION

 CAS NO.
 100-48-1

 EINECS NO.
 202-856-2

 FORMULA
 (C₅H₄N)CN

 MOL WT.
 104.11

 H.S. CODE
 2933.33

TOXICITY Oral rat LD50: 1185 mg/kg

SYNONYMS Isonicotinonitrile; 4-Pyridinecarbonitrile;

4-Azabenzonitrile; Nicontinic acid nitrile; 4-Pyridinenitrile;

DERIVATION

CLASSIFICATION

PHYSICAL AND CHEMICAL PROPERTIES

PHYSICAL STATE white to off-white crystalline powder

MELTING POINT 77 - 78 C BOILING POINT 194 - 916 C

SPECIFIC GRAVITY

SOLUBILITY IN WATER 40 g/l

рН

VAPOR DENSITY AUTOIGNITION

NFPA RATINGS Health: 2; Flammability: 1; Reactivity: 0

REFRACTIVE INDEX

FLASH POINT 84 C

Stable under ordinary conditions

GENERAL DESCRIPTION & APPLICATIONS

PYRIDINE, also called azabenzene and azine, is a heterocyclic aromatic tertiary amine characterized by a six-membered ring structure composed of five carbon atoms and a nitrogen which replace one carbon-hydrogen unit in the benzene ring (C₅H₅N). The simplest member of the pyridine family is pyridine itself. It is colorless, flammable, toxic liquid with a unpleasant odor, miscible with water and with most organic solvents, boils at 115 C. Its aqueous solution is slightly alkaline. Its conjugate acid is called pyridinium cation, CsHsNH+, used as a oxidation agent for organic synthesis.. Pyridine is a base with chemical properties similar to tertiary amines. Nitrogen in the ring system has an equatorial lone pair of electrons, that does not participate in the aromatic pibond. Its aqueous solution is slightly alkaline. It is incompatible and reactive with strong oxidizers and strong acids, and reacts violently with chlorosulfonic acid, maleic anhydride, oleum, perchromates, b-propiolactone, formamide, chromium trioxide, and sulfuric acid. Liquid pyridine easily evaporates into the air. If it is released to the air, it may take several months to years until it breaks down into other compounds. Usually, pyridine is derived from coal tar or synthesized from other chemicals, mainly acetaldehyde and ammonia. Pyridine compounds are found in nature. For example, nicotine from tobacco, ricinine from castor bean, pyridoxine or vitamin B and P products, alkaloids (such as coniine, piperine and nicotine), and etc. Some pyridine compounds consumed largely are;

Picoline: Three structural isomers of methyl pyridines (alpha, beta, gamma-positions)
Lutidine: Six structural isomers of dimethyl pyridines (2,3-, ,24-, 2,5-, 2,6-, 3,4-, 3,5- positions)
Collidine: Three structural isomers of trimethyl pyridines (2,3,5-, 2,3,6-, 2,4,6- positions)

Pyrimidine: Pyridine alteration containing nitrogen atoms at positions 1 and 3

Piperidine: Hexahydropyridine (saturated form) Nicotinic acid: pyridine-3-carboxylic acid

Pyridine and its derivatives are very important in industrial field as well as in bio chemistry. Nucleotide consist of either a nitrogenous heterocyclic base (purine or pyrimidine). Three major pyrimidines in living systems are cytosine, thymine, and uracil. Pyrimidine and its derivatives are biologically important components of nucleic acids (DNA, RNA) and coenzymes. Some pyridine system is active in the metabolism in the body. Certain nitrogenous plant products also have pyridine class compounds. They can be the parent compound of many drugs, including the barbiturates. Pyridine and its derivatives are used as solvents and starting material for the synthesis of target compounds such as insecticides, herbicides, medicines, vitamins, food flavorings, feed additives, dyes, rubber chemicals, explosives, disinfectants, and adhesives. Pyridine is also used as a denaturant for antifreeze mixtures, as a dyeing assistant in textiles and in fungicides. Cyanopyridines, nicotinonitriles, are intermediates for the preparation of pharmaceuticals and agrochemicals.

SALES SPECIFICATION

APPEARANCE white to off-white crystalline powder

ASSAY 98.0% min WATER 0.5% max

TRANSPORTATION

PACKING

HAZARD CLASS 6.1 (Packing Group: III)

UN NO. 2811

OTHER INFORMATION

Hazard Symbols: XN, Risk Phrases: 20/21/22, Safety Phrases: 24/25